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**EVALUATION OF PROXIMATE COMPOSITION AND SELECTED
PHYSICO-CHEMICAL PROPERTIES OF BUTTERNUT SQUASH**

(CUCUBITA MOSCHATA)

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(*CUCUBITA MOSCHATA*)**

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Abstract

Purpose: This study sought to evaluate of proximate composition and selected physico-chemical properties of butternut squash.

Methodology: Samples were chosen randomly from Northern part Ghana specifically Kukobila and data analyzed statistically. Oil was extracted using soxhlet extraction method.

Findings: Results showed that For the proximate analysis, fat content of the pulp was 1.16% and carbohydrate 72.70%. The fibre content of the defatted seed residue was 5.59%, ash (6.86%) and protein (49.20%). The acid value, saponification value, iodine value, peroxide value and free fatty acid were 3.42mgKOH/g, 15.43mgKOH/g, 21.08gI₂/100g, 2.50meqO₂/kg and 1.71 respectively. The high oil content of the seeds coupled with fairly low acid value makes it suitable as edible oil, a food supplement and suggest its suitability for the production of paints, inks but not suitable for soap and biodiesel production

Unique contribution to theory, practice and policy: Farmers should avoid local consumption of butternut squash due to lack of knowledge on the potentials and uses; since its underutilized and cultivated mainly for export. More research should be carried out to convert Butternut squash into breakfast meals and baby feeds due to its high protein content and gluten free nature

Keywords: *Pulp, nutritive, defatted, biodiesel, fibre, formulations, Cucurbita moschata, proximate composition, oil physicochemical properties*

INTRODUCTION

The current rate of population growth coupled with climate change leaves the world with little food to feed the masses; there is therefore the urgent need to identify and make use of neglected and under-utilized plant species for food production. Butternut squash is the most popular vegetable among the winter squash varieties. Butternuts are annual long trailing vines, usually cultivated in warmer climates of South and Central American. They produce edible fruits, flowers and seeds⁵⁵.

It is cultivated in most tropical countries in Africa especially in Southern Africa. The fruit is popular for making pumpkin pie, sweets and desserts such as; steamed food flesh with granted coconut and sugar. In Zambia, ripe butternut flesh is dried for longer preservation. About 40% of the households in Zambia use butternut leaves as relish daily during the rainy season. In Cameroon and other parts of Central and West Africa, butternut squash is principally grown for the ripe seeds. The seeds are roasted with its shells, salted and eaten as snacks. The seed oil is edible and used as fuel. It has several medical applications in Thailand and China. Crushed fresh seeds are used as anthelmintic, and also used in treatment of skin infections and inflammations²⁷.

In Ghana, butternut has recently been cultivated in the Upper East, Upper West and Northern Regions mainly for export. According to Ministry of Food and Agriculture (MOFA) butternut squash cultivation is a strategy to increase income and reduce poverty in Northern Ghana especially. It has now become an alternative crop for Volta farmers' too⁴⁹. However, the local consumption and the use of this plant have not been encouraged until now. Butternut squash is a vigorous growing plant that takes 85 to 90 days to mature and it produces good uniform fruits weighing about 650 to 1000 gm. Among other cucurbits like courgettes, pumpkin and cucumbers, butternut squash is gaining preference because of its early maturity, its color that is appealing to the eyes of the customers, its long shelf life and small size that makes it attractive to consumer.

Moreover, it is a hardy crop, which is able to tolerate moderately harsh environmental conditions, resistant to many pests of cucurbits and can be grown well in small plots of land. It yields relatively good returns with minimal fertilizer inputs making it an appropriate crop as land subdivision continues in rural settings. The crop also has ready local and export market⁶ some growers in Western region of Kenya are expanding their production to meet increased local and export demand³⁶. Although butternut was introduced recently in Ghana, most farmers are going into its cultivation but the local consumption has not caught up well in Ghana; as many people including the farmers make very little or no use of the produce. The produce are mainly exported, therefore there is the need to encourage local consumption. This will however be possible if adequate research and development is done to assess potential uses.

MATERIALS AND METHOD

Sampling Collection

Butternut fruits were sampled from the local farm at Kukobila near Tamale in the Northern region of Ghana. The fruits were chosen randomly at commercial maturity according to their similarity in color, size and the absence of surface defects.

Sample preparation

The butternut fruit (fig. 1) were peeled and the seeds (fig. 2) were extracted manually from the pulp. The flesh pulp was sliced into thin sizes and dried in the sun to remove all the moisture in it. The dry samples were grinded into powder and stored in a clean dry container until ready

for use. The seeds were also dried for one week. The immature or broken seeds and foreign materials were removed and the rest stored in a clean air tight container in the dark until needed. The seeds were grinded in a grinder mill to produce the powdered form.



Figure 1: Photos of Butternut (*Cucurbita moschata*).



Figure 2: Photos of Butternut Seeds.

Methods for preparing flour

Cleaning and washing: Butternut squash fruits were washed with clean water in order to remove dirt and any unwanted materials.

Peeling: The fruits were peeled using a sharp knife due to its hard skin.

Drying: The fruits were sliced into two, seeds were removed and the fresh pulp was cut into thin sizes and dried in the sun for about one week.

Grinding: The dried pulp was grinded using a grinding mill into flour.

Proximate Analysis of fruit pulp, defatted seed residue and seeds

Moisture, ash (AOAC method No. 945.18), fat (AOAC method No. 945.16) and crude protein (AOAC method No.920.53) were determined by the AOAC⁴. A factor of 6.25 was used for the conversion of nitrogen into crude protein. Carbohydrate content was calculated by subtracting the contents of all the measured components (moisture, ash, crude fat and crude protein) from total mass. All measurements were expressed on dry matter basis and reported as a mean (\pm S.D) of triplicate analysis. Determination of crude fat or lipid content of the samples was done using Soxhlet (Cehmglass) type of the direct solvent extraction using petroleum ether (boiling range 60-80^oC) as solvent.

Moisture Content Analysis

A mass of 3.0 g of each sample was weighed into a dried and pre-weighed glass crucible (W1). Samples were dried in an oven at 105.2 °C overnight. The crucible was cooled at room temperature and weight taken as (W2).The moisture content was calculated using the formula;

$$\% \text{ moisture} = \frac{W1 - W2}{\text{weight of sample}} \times 100 \%$$

Crude Ash Analysis

A Cleaned empty crucible was placed in a muffle furnace for 30 minutes at 600 °C, cooled and weighted (W1). A mass of 1.0 g of each sample was put in the crucible and weight noted (W2). The sample was ignited in the furnace at 600 °C for 3 hours. The appearance of grey white ash indicated complete oxidation of all organic matter in the samples. After ashing, the crucible was cooled and weighed (W3). The following formula was used to calculate the percentage ash.

$$\% \text{ ash} = \frac{W3 - W1}{\text{weight of sample}} \times 100 \%$$

Crude Fiber

A mass of 2.0 g of the defatted samples was weighed and transferred into porous crucibles. Crucibles were placed into a Dosi- fiber unit and the valve at off position. A volume of 1.50 ml of pre-heated H₂SO₄ solution and some drops of acetone were added to prevent foaming. The cooling circuit and the heating element (power at 90 %) were turned on. Power was reduced to 30 % when it started boiling and left for 10 minutes. The valves were opened to allow drainage of acid and rinsed with distilled water to completely remove all acid from the sample. The sample was dried in an oven at 150 °C for 1 hour and then at room temperature and weighed (W1). The Samples were then kept in a furnace at 55 °C for 3-4 hours. The samples were cooled and weighed again (W2). The crude fiber was calculated using the formula:

$$\% \text{ Crude fibre} = \frac{W1 - W2}{\text{weight of sample}} \times 100 \%$$

Lipid

Lipid content was estimated using TecatorSoxtec. A quantity of 2 g sample mixed was weighed into a thimble and covered with absorbent cotton, while 40 ml of petroleum ether (40-60 °C Bpt) was pre-weighed cup. Both thimble and cup were attached to the extraction unit. The sample was extracted using ethanol for 30 minutes and rinsed for 50 minutes. The solvent was evaporated from the cup to the condensing column. Extracted fat (Appendix-VIII) in the cup was then place in an oven at 105^oC for 1h and cooled and weighed.

Percent fat was calculated as:

$$\% \text{ Lipid} = \frac{[\text{initial Cup weight} - \text{Final Cup weight}]}{\text{weight of Sample}} \times 100$$

Crude protein Analysis

Determination of total nitrogen (crude protein) was conducted using the Kjeldahl method. The sample was digested in sulfuric acid using CuSO_4 as a catalyst. N was converted into NH_3 , then distilled, trapped in boric acid and titrated with H_2SO_4 .

A mass of 1 g of the sample was weighed into a digestion tube (250 ml). Two tablets (CuSO_4) were added as a catalyst. A volume of 13 ml of concentrated H_2SO_4 was added and inserted into a rack. The mixture of standard sample was placed in a digestion block heater under fume hood, and the exhaust manipulation fold was connected to an aspirator. The mixture was kept in a digester at 420°C until the liquid becomes transparent. The rack was removed from exhaust manifold and transferred separately into distillation unit. Distill automatically by adding 65 ml distilled water and 35 ml of 40 % sodium hydroxide solution. The condensed liquid was collected in a cornical flask containing boric acid and titrated against 0.1142N sulfuric acid until color turns purple. The crude protein was calculated as;

$$\% N = \frac{1.4007 \times (T-B)}{W} \times N$$

Where;

T= volume of acid used for sample titration

B = volume of acid used for blank

N = normality of acid

W = sample weight in grams

1.4007: conversion factor milliequivalent of nitrogen and N percent.

% Crude protein can be calculated as;

$$\% \text{CP} = \% \text{N} \times 6.25$$

Where 6.5 is a dilution factor for all forages.

Carbohydrate

The carbohydrate content was determined by difference, that is, addition of fat, crude protein, ash, and crude fiber subtracted from 100%. This gives the amount of nitrogen-free extract otherwise known as carbohydrate.

$$\% \text{CHO} = 100 - (\% \text{moisture} + \% \text{Fat} + \% \text{Ash} + \% \text{Crude fiber} + \% \text{Crude protein})$$

Energy level.

The energy level was calculated using the formula;

$$\text{Energy (Kcal)} = [(\% \text{CHO} \times 4) + (\% \text{CP}) + (\% \text{Fat} + 9)]$$

Where; CHO, CP and CL stand for carbohydrate, crude protein and crude lipid respectively.

OIL EXTRACTION

The method employed was that of solvent extraction with the Soxhlet extractor apparatus as described by AOAC 1990.

A 250 ml Soxhlet extractor apparatus and petroleum ether as solvent were used. A mass of 30 g of butternut seeds powder was weighed into a muslin cloth which was placed in a Soxhlet

apparatus thimble. A round bottom flask containing 250 ml of petroleum ether was to the end of the apparatus and a condenser, tightly fixed at the bottom the extractor. The whole set up was heated up in a water bath at a temperature of 60 °C. The excess solvent in the oil was recycled by heating in a heating mantle at a temperature of 60 °C after the extraction. Quantity of oil extracted was determined gravimetrically. The oil yield (fig 3) was evaluated as the ratio of the weight of the extracted seed oil to the weight of the butternut powder sample (Appendix-I) as described below.

$$\% \text{ Oil yield} = \frac{\text{Weight in gram of extracted oil}}{\text{Weight in gram of extracted seed powder sample}}$$

The extraction process was carried out for six hours and was done in correspondence to the solvent used (petroleum ether at 60 °C).



Figure 3: Butternut squash seed oil.

Physico-chemical properties

Physical properties

Color and Odor

The color of oils describes the physical appearance the oil by sensory evaluation as compared to rainbow Color and the Odor is its smell.

These were done by direct physical observation of the oil and compared to other literature.

pH

This is the degree of acidity or alkalinity of the oil and it was measured with the pH meter.

The pH electrode was standardized with buffer solution and the electrode immersed into the oil sample. pH and temperature readings displayed on the pH meter screen were noted.

The experiment was repeated and the average value was taken to be the pH of the oil at that temperature

Determination of Specific Gravity

A beaker was placed on a chemical balance and a volume of oil was weighed.

The same volume of distilled water was weighed using another beaker.

The specific gravity was calculated as,

$$SG = \frac{\text{Measured weight of oil}}{\text{Measured weight of distilled water}}$$

3.5.1.4 Refractive Index

The refract meter's prism case was opened and four drops of the oil were placed on the bottom of the prism using a plastic pipette.

Visual observations were made the instrument's eyepiece and the prism adjustment was turned until a distinct light or dark border became visible.

The compensator dial was adjusted to remove as much Color as possible from both the light and dark areas. The eyepiece was also adjusted to bring the light or dark border into a sharp focus while adjusting the prism knob to place border in the center of the crosshairs.

The refractive index and the temperature of the oil were read from the scale of the instrument.

Chemical properties

Acid value

A mass of 1.0 g was completely dissolved in 20 ml of ethanol

Using phenolphthalein as indicator, the mixture was titrated against 0.1 M KOH Solution until a pink Color appeared and persisted for at least 10 seconds.

The experiment was repeated and the average of the titer values was taken.

The acid value was calculated as;

$$AV = 56.1(VN)/M$$

Where;

V= Titer value of KOH used

N= Normality of the KOH used

M= mass of oil used

Free Fatty Acid Value

The Free Fatty Acid value was calculated as;

$$\text{Acid Value} \times 0.503^{28}$$

Peroxide Value

A mass of 5.0 g of the oil was weighed into a conical flask. A volume of 10 ml of chloroform was added and swirled to dissolve the oil.

A volume of 15.0 ml acetic acid and 1ml KI solution was added to the content, shaken and left in the dark place for exactly 5 minutes.30 ml of distilled water was added to the content and the titrated against excess sodium thiosulphate solution. The endpoint was reached when the blue color disappeared.

The experiment was repeated without the oil under the same conditions and was used as the control.

The peroxide value was calculated as;

$$PV = 100N (VS - VC) \div M$$

Where;

VS= Volume of Na₂S₂O₃ used in the sample titration

VC= Volume of Na₂S₂O₃ used in control titration

N= Normality of $\text{Na}_2\text{S}_2\text{O}_3$

M= Mass of oil sample

Saponification Value

Indicator method was used as specified by ISO 3657(1988). A mass of 2 g of the sample was weighed into a conical flask; 25 ml of 0.1 N ethanolic potassium hydroxide was added. The content was constantly stirred and allowed to boil gently for about 60 minutes. A reflux condenser was placed on the flask containing the mixture. Few drops of phenolphthalein indicator was added to the warm solution and then titrated with 0.1 M HCl to the end point until a pink of the indicator disappears. The same procedure was used for other samples and the blank.

The expression for saponification value (S.V) (Appendix-VI) is given by;

$$S.V = 56.1 N(V_0 - V_1) \div M$$

Where; V_0 is the volume of the solution used for the blank test, V_1 is the sodium thiosulphate used for the determination, N is the actual Molarity of HCl used and M is mass of sample.

Iodine Value

The iodine value was determined according to the methods of ISO 3961:1996.

Approximately 0.25 g of the oil sample was weighed into a 250 ml conical flask. 10 ml of chloroform was added. 30 ml of Hanus solution was added and closed by para film. The mixture was allowed to stand for 30 minutes with continuous shaking. 10 ml of 15% potassium iodide solution was added and then shook. 100 ml of distilled water was then added. The resulting solution was titrated against 0.1 N sodium thiosulphate solution till yellow color formed, then 2-3 drops of starch solution was added where blue solution formed and then continued with the titration until the blue color disappeared (volume of $\text{Na}_2\text{S}_2\text{O}_3$ at the end point represents S). The same above procedure but without sample was done (blank).

The iodine number was calculated by using the formula:

$$\text{Iodine value} = \frac{(B-S) \times N \text{ of Na}_2\text{S}_2\text{O}_3 \times 0.127 \text{g/meq}}{\text{Weight of Sample}} \times 100$$

B= Vml of $\text{Na}_2\text{S}_2\text{O}_3$ volume for blank

S= Vml of $\text{Na}_2\text{S}_2\text{O}_3$ volume for sample

RESULTS

The tables below signify the proximate composition of butternut squash fruits and the physico-chemical properties of the seed oil.

Table 3

Proximate Composition of Dried Pulp, Dried Seed and Defatted Seed Residue.

Component Analyzed	Dried Pulp	Dried Seed	Defatted Seed Residue
Moisture Content (%)	6.17 ± 0.50	4.00 ± 0.50	5.16 ± 0.02
Liquid Content (%)	1.16 ± 0.16	22.83 ± 0.17	4.33 ± 1.88
Ash Content (%)	5.57 ± 0.27	4.60 ± 0.30	6.86 ± 0.05
Fiber Content (%)	3.55 ± 0.30	1.152 ± 0.20	5.59 ± 0.10
Crude Protein Content (%)	10.85 ± 0.01	33.92 ± 0.25	49.2 ± 0.10
Carbohydrate Content (%)	72.70 ± 0.00	33.50 ± 0.00	28.86 ± 0.00

Values are means of two determinations with SD ± two decimal places

Table 4

Physico-Chemical Properties of Extracted Oil and Oils Found in Literature.

Physico-chemical Property	Oil Source (seed)				
	* <i>C. pepo</i>	* <i>C. maxma</i>	*Desert melon	*Water melon	<i>C. moschata</i>
Color	NR	NR	NR	L. yellow	L. brown
Odor	NR	NR	NR	NR	Pleasant
pH	NR	NR	NR	NR	6.91 ± 0.01
RI	1.47	1.46	1.47	1.47	1.37 ± 0.00
OC %	41.59	43.69	28.00	41.32	13.31 ± 0.04
AV mgKOH/g	0.78***	0.53	2.50	2.37	3.42 ± 0.01
SV mgKOH/g	190.69	185.20	182.20	183.13	15.43 ± 0.07
V mm ² /s	93.65	48.09	NR	2.48	NR
IV gI ₂ /100g	104.36	105.12	124.00	121.51	21.08 ± 0.49
FFA	0.39***	0.27	NR	6.40	1.71 ± 0.00
SG	0.91	0.91	0.95	0.85	0.99 ± 0.00
PV meqO ₂ /kg	NR	NR	NR	2.20	2.50 ± 0.02

**Mean standard deviations in two dimensions.

AV=Acid Value, SV=Saponification Value, IV=Iodine Value, V=Viscosity, FFA=Free Fatty Acid, NR=Not reported, OC = % Oil yield,* Are oils found in literature, ***²⁶.

DISCUSSIONS

The difference in the moisture content of the dried seed and the defatted seed is as a result of high concentration of defatted seed which is similar to the whole seed (sun dried) of *C.*

moschata (5.7 ± 0.4), seed kernel (sun dried) of *C.moschata* (5.6 ± 0.0), *C. maxima* whole seed (sun dried) (5.5 ± 0.2)²². The moisture values are lower than that of soya bean (11.07%) and coconut seeds, (14.3%)²¹. The low level of moisture content of butternut squash seeds enables them to be preserved for long period of time. Moisture levels of food products have a bearing on their dry matter content. The higher the moisture content, the lower the dry matter content; which would help the food processors to make decisions on the economics of thermal processing²². The crude protein of processed samples of fruit and seeds in this study are lower than that of *C.moschata* fruit without rind (4.3 ± 0.0), *C. maxima* (3.7 ± 0.0), and whole seed of *C. moschata* (sun dried) (35.4 ± 0.3), seed kernel of *C.moschata* (37.4 ± 0.2), whole seed of *C. maxima* (sun dried) (36.3 ± 0.2) and seed kernel of *C. maxima* (40.3 ± 0.3)²². A difference exists between the dried seed and defatted seed residue which could be as a result of sample preparation method. The protein content of the defatted seed residue could contribute to the protein content in the seed's kernel²². An adult male of about 70 kg body weight requires 35 g of protein daily, therefore, only 98.93 g of *C. moschata* would be required to provide minimum daily protein needs. However, 122 g should be consumed to meet requirements if an allowance of 25 % is made to take care of digestibility and limiting sulphur amino acid²⁰. The proteins are globulin type and are deficient in lysine and sulphur bearing amino acids⁵⁷. The results shows that *C.moschata* seeds from Kukobila in the northern part of Ghana are rich in proteins hence good for children, lactating mothers and old people who need more proteins for growth, maintenance and repair of worn tissues. The crude lipid content of dried seed was within the range of 20 – 38.00 %. This could be classified as an oil seed like groundnut, melon etc. Lipids are essential because they provide the body with maximum energy⁴⁵.

The fiber content of the defatted seed residue was higher than that of the powdered seed which could be as a result of the husk of the seed. The fiber content of the dried seed is 1.15 % which is lower than that of melon (2.7 %) ⁴³ but can be compared favorably with that of *Cucurbita pepo L* seeds (1.0)¹⁸. Fiber containing foods are known to expand the inside walls of the colon, ease the passage of waste by preventing constipation. It lowers the cholesterol levels in the blood and reduces the risk of various cancers. Emphasis is made on the intake of low fiber in the nutrition of infants and weaning children; high fiber levels in weaning diet can lead to irritation of gut mucosa¹². It also enhances gut perturbation in young animals (e.g. piglets and chicken)¹⁹.

The Ash content of the dried seeds are similar to that of *C.pepo L* seed (5.50)¹⁸ melon (5.40%) and the defatted seed residue similar to melon (7.9%)⁴³. Samples with high ash content are expected to have high concentration of various mineral elements which are expected to speed up metabolic processes, improve growth and development. The carbohydrate content of the dried seed and defatted residue cannot be considered as a potential source of carbohydrate when compared to dried pulp (72.70 %) and the content of some convectional sources like cereals with 72.90 g/100g carbohydrates³.

The experimental results in Table 1 showed that butternut squash fruit cultivated at Kukobila has seeds of immerse values. The extracted oil was light brown in color with an average yield of 131g of the prepared sample. The percentage yield of the extracted butternut squash seed oil was 13.31 % (Table 1). This value fell in the range reported for different species of *cucurbita* (9.8-52.1 %) ²⁶. In comparison to the oil content of other seeds, the oil content of this study was lower than that of *C. maxima* (43.69 %), *C. pepo* (41.59 %) and desert melon (28.0). The difference in oil yield may be as a result of geographical location, specie or variety of fruit, method for sample preparation, size of fruit (in terms of maturity) and other environmental factors. This can be as a result of the husk. The color and odor of the oil is an important feature

which determines the customers' acceptability for oil products. Butternut squash seed oil has a light brown color with a pleasant smell. The color development is due to the presence of pigment such as chlorophyll, carotenoids etc. Removal of such coloring pigment is an important task from commercial view point. Bleaching can be used to remove the color of oils and fats during processing¹².

The refractive index of *C.moschata* was 1.3692 which is closer to that of pumpkin seed oil; *C.pepo* (1.4662), *C.maxima* (1.4652) and desert melon (1.468). It is independent on the variety of pumpkin seed oil and it's not as thick as most drying oils whose refractive indices fall between 1.475-1.485⁵. Correspondingly the effect of variety was not significant on the specific gravity of pumpkin seed oil. The specific gravity (0.997) in this study does not differ much from *C.maxima* (0.913), *C.pepo* (0.915) and Desert melon (0.954).

The saponification value of *Cucurbita moschata* was (15.43) which is far lower than that of *C. pepo* (190.69), *C. maxima* (185.20) and desert melon (182.1). This indicates that butternut squash seeds oil is not a good raw material for soap industries. Saponification value is inversely proportional to the average molecular weight or chain length of the fatty acid³⁸. Therefore, the shorter the average chain length ($C_4 - C_{12}$) the higher the saponification number⁵¹. The value obtained from this study shows that it contains high amounts of long chain fatty acids ($> C_{12}$), which is lower than the regulation of codex standard permissibility level (CODEX-STAN210-1999) (190-209 mg KOH/g).]The Acid value (0.34) of *Cucurbita moschata* was lower than that of *C. pepo* (0.39), *C. maxim* (0.53) and desert melon (2.5) indicates that *cucurbita moschata* contains less Free Fatty acid thus reducing its susceptibility to rancidification or deterioration. It is a measure of the extent to which triglycerides in the have been decomposed by lipase action into free fatty acids; acid value depends on the degree of rancidity which is used as an index of freshness⁴².The low acid value suggest that the oil is suitable for paint and ink production and edible as well. The low free fatty acid value (1.71) indicates that butternut squash seed oil is of good quality².

The peroxide value of the extracted oil was 2.5 meq $O_2 kg^{-1}$; indicating good oxidative stability of the oil. Comparing with that of watermelon seed oil (2. 2 meq $O_2 kg^{-1}$), butternut squash seed oil is higher than watermelon seed oil²⁸. Peroxide value is as a result of changes the fat undergo during storage; emission of unpleasant odor and taste (rancidity). The results obtained from the study shows that butternut squash seed oil is fresh and does not easily undergo rancidification during storage⁴⁶ oils with peroxide values ranging between 20.0-40.0 easily undergoes rancidification⁴.The iodine value of the extracted oil was 21.082 mg $I_2/ 100g$ which is lower than *C.pepo* (104.36), *C.maxima* (105.12) and desert melon (124.0). This result suggests that butternut squash seed oil is not suitable for biodiesel production when compared with the other varieties of pumpkin seed oil. Low iodine value of vegetable oil produces biodiesel with high cloud and pour points; higher cloud and pour points means poor engine performance at cold temperatures. Similarly, oils with iodine value less than 100 g $I_2/100g$ are non-drying oils and have lesser number of unsaturated bonds; lower susceptibility to oxidative rancidity. This non-drying attribute is a good fuel property which indicates the oil's resourcefulness as feedstock for biofuel production.

CONCLUSION

The high oil content of the seed of butternut squash coupled with fairly low acid value makes it suitable as edible oil, a food complement and suggests its suitability for the production of paints, inks and others. Butternut squash seed oil is not suitable for soap and biodiesel production; due the low iodine and saponification values. The seeds are highly nutritious; rich

in protein which is good for children, lactating mothers and old people who need more protein for growth, maintenance and repair of worn tissues.

RECOMMENDATION

In order to prevent post-harvest losses, butternut squash fruits which are rejected before exportation can be used to prepare animal feed or used as a meal supplement for livestock. Further research should be carried out to convert Butternut squash into breakfast meals and baby feeds due to its high protein content and gluten free nature

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